TOTAL SYNTHESES OF (±)-TELEOCIDIN B-3 AND B-4<sup>1)</sup>

Shin-ichi Nakatsuka,\* Toshiya Masuda, and Toshio Goto

Laboratory of Organic Chemistry, Faculty of Agriculture, Nagoya University, Chikusa, Nagoya 464, Japan

Abstract; First total syntheses of  $(\pm)$ -Teleocidin B-3(1c) and B-4(1d) were achieved by several functionalizing reactions starting from indole.

Teleocidin B was first isolated as a strong skin-irritant compound from the mycelium of <u>Streptomyces mecliocidicus</u> by Takashima and Sakai.<sup>2)</sup> Recently, Fujiki <u>et al.<sup>4)</sup></u> reported that teleocidin B is one of the most potent tumor promoters known. The chemical structure was determined by Hirata's group<sup>3)</sup> by chemical methods and X-ray analysis of the monobromoacetate of its dihydro derivative in 1966. Fujiki and Sakai's group<sup>5)</sup> found that the compound was a mixture of four stereoisomers [B-1(1a), B-2(1b), B-3(1c), and B-4(1d)] and that Hirata's teleocidin B derivative used for X-ray analysis was derived from teleocidin B-4. Teleocidin B is a unique natural product having an amino group at the 4-position and a modified mono terpenyl moiety at the 6,7-positions of the indole nucleus.

We have developed new functionalizing methods<sup>6,7)</sup> for the benzene part of indole derivatives and applied them to the efficient syntheses of indolactam  $V^{8)}$  (the minimum structure for biological activity of teleocidin) and an indole derivative<sup>9)</sup> containing the same substituents at the 6,7-positions of teleocidin B. Now we report the first total syntheses<sup>10)</sup> of teleocidin B-3 (1c) and B-4(1d).

We prepared benzoate  $3^{9}$  by radical oxidation of 2 with benzoyl peroxide in 30% yield. Selective nitration<sup>11)</sup> of 3 was achieved by treatment with AcONO<sub>2</sub> in acetic anhydride. Other nitration conditions (for instance, conc. HNO<sub>3</sub> in acetic acid) could not be used because of instability of the product in such



Teleocidin B (1) [ mixture of B-1(1a), B-2(1b), B-3(1c), and B-4(1d) ]

acidic conditions. The position of the nitro group was established by comparison of the chemical shift of the 5-proton with those of the other 4-nitro derivatives.<sup>11)</sup> The benzoyl group of the nitro compound was selectively hydrolyzed with 1 N NaOH to give  $4^{12}$  in 60% overall yield from 3. Oxidation of 4 with MnO<sub>2</sub> in CH<sub>2</sub>Cl<sub>2</sub> at 25°C for 1.5 hr afforded a lactam  $5^{12}$  in 74% yield. Although the lactam 5 was a mixture of trans- and cis-isomers (5a:5b = 5:1), the former (5a), mp 178.5-180 °C, was obtained in pure form by crystallization.

Interestingly, hydrolysis of 5a with 6 N NaOH at 60°C was accompanied by decarboxylation and subsequent reduction of the product with BH<sub>3</sub> in THF at 25 °C for 20 min afforded an alcohol  $6a^{12}$  in 66% yield. Compound 6a was treated with formalin and dimethylamine in acetic acid at 60°C for 15 min and followed by treatment with methyl nitroacetate in the presence of Et<sub>3</sub>N in chlorobenzene at 100°C for 10 min to give 7 (95%, 1:1 mixture of stereoisomers).

Selective reduction of the 4-nitro group of  $\chi$  by catalytic hydrogenation was unsuccessful, unlike our indolactam V synthesis,<sup>8</sup>) but reduction with sodium dithionite in the presence of Et<sub>3</sub>N in MeOH-H<sub>2</sub>O gave the corresponding 4-amino compound & in 52% yield. Introduction of a C<sub>5</sub> unit to & was achieved with 2-oxoisovaleric acid in DMF and subsequent reduction with NaBH<sub>3</sub>CN to give 9. Compound 9 was a mixture of four stereoisomers and could be easily separated on a silica gel TLC plate into two mixtures containing two stereoisomers,  $9a^{12}$  (35%, 1:1 mixture) and  $9b^{12}$  (26%, 1:1 mixture). Presumably these isomers are epimers related to the stereochemistry between the valinyl moiety and the terpenyl substituent at the 6,7-position because of the interaction of those substituents on the very flat tetrahydrobenzindole system. These stereochemistries were confirmed by the subsequent derivations to teleocidin B-3 and B-4.

Compound 9a gave a lactam  $10a^{12}$  in 29% yield by the treatment with Nhydroxysuccinimide (HOSU) and dicyclohexylcarbodiimide (DCC) in CH<sub>3</sub>CN at 25°C for 30 min followed by H<sub>2</sub>/PtO<sub>2</sub> in methanol for 15 min. The epimer of 10awas not isolated.

Transformation of the hydroxyethyl group of 10a to the terminal olefin found in teleocidin B was a very difficult process because this alcohol cyclized<sup>13)</sup> easily at the 1-position of the indole nucleus. We chose the cis-elimination method of selenoxide derivative. Treatment of 10a with MsCl in pyridine followed by NaSePh in ethanol to give a selenide. Reduction of the ester group of the selenide with NaBH<sub>4</sub>/LiCl in THF-EtOH gave 11a which was isolated on silica gel TLC in 48% yield. Oxidation of the phenylselenenyl group of 11a with KIO<sub>4</sub> in THF-MeOH followed by heating at 60°C gave terminal olefin 12a<sup>12</sup>) in 59% yield.

Methylation<sup>3g)</sup> of 12a was achieved with CH<sub>3</sub>I/NaHCO<sub>3</sub>-Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> in EtOH at 70 °C in a sealed tube for 4.5 hr to give teleocidin B-4 (1d) in 44% yield (54% of 12a was recovered). In the same manner, we also prepared teleocidin B-3

CO<sub>2</sub>CH<sub>3</sub>



2 : X=H 3 : X=OBz









cis isomer

5Ь

9a ≫ 9b 26%





11a



: R=H 12a đ Teleocidin B-4: R=CH3 ۱d 44%



a)  $AcONO_2$  b) NaOH c)  $MnO_2$  d)  $BH_3THF$  e)  $CH_2=N^+(CH_3)_2(OAc)^-$  f)  $O_2NCH_2CO_2CH_3$ h)  $(CH_3)_2CHCOCO_2H/NaBH_3CN_i)$  HOSu / DCC j)  $H_2$  / PtO<sub>2</sub> k)  $/ \text{Et}_{3}^{N}$  g)  $\text{Na}_{2}^{S}_{2}^{O}_{4}$ MsCl/Py 1) NaSePh m) NaBH4-LiCl o) KIO4 p) 60 °C q) CH3I

(1c) from 9b in 4.5% overall yield. <sup>1</sup>H-NMR, MS, and UV spectra and retention times on HPLC of the synthetic  $(\pm)$ -teleocidin B-3 (1c) and B-4 (1d) were completely identical with those of natural teleocidin B-3 and B-4, respectively.

ACKNOWLEDGEMENT We thank Dr. Hirota Fujiki in National Cancer Research Institute, Tokyo, for generous gift of natural teleocidin B-3 and B-4 and  $^{1}$ H-NMR spectral data. This work was supported by the Grants-in-Aid for Cancer Research No. 61010093 from the Ministry of Education, Science and Culture.

## REFERENCES AND FOOTNOTES

- 1. Synthetic studies on teleocidin V.8)
- 2. M. Takashima and H. Sakai, Agric. Biol. Chem., 24, 647(1960).
- 3. H. Nakata, H. Harada, and Y. Hirata, Tetrahedron Lett., 2125(1966). H. Harada, N. Sakabe, Y. Hirata, Y. Tomiie, and I. Nitta, Bull. Chem. Soc. Japan, <u>39</u>, 1773(1966). Japan, <u>39</u>, 1773(1966).
  Related natural products were reported recently. a) Teleocidin A.<sup>2</sup>
  b) Lyngbyatoxin A: J. H. Cardellina II, F. -J. Marner, and R. E. Moore, Science, <u>204</u>, 193(1979). c) Indolactam V: K. Irie, M. Hirota, N. Hagiwara, K. Koshimizu, H. Hayashi, S. Murao, H. Tokuda, and Y. Ito, Agric. Biol. Chem., <u>48</u>, 1269(1984). d) Olivoretin A and D: S. Sakai, N. Aimi, K. Yamaguchi, Y. Hitotsuyanagi, C. Watanabe, K. Yokose, Y. Koyama, K. Shudo, and A. Itai, Chem. Pharm. Bull., <u>32</u>, 354(1984). e) Olivoretin B and C: Y. Hitotsuyanagi, K. Yamaguchi, K. Ogata, N. Aimi, S. Sakai, Y. Koyama, Y. Endo, K. Shudo, A. Itai, and Y. Iitaka, Chem. Pharm. Bull., <u>32</u>, 3774(1984). f) Blastmycetin A-C: K. Irie, N. Hagiwara, T. Kurome, H. Hayashi, M. Arai, and K. Koshimizu, Agric. Biol. Chem. <u>51</u>, 285(1987). g) Olivoretin E: S. Sakai, Y. Hitotsuyanagi, K. Yamaguchi, N. Aimi, K. Ogata, T. Kuramochi, H. Seki, R. Hata, H. Fujiki, M. Suganuma, T. Sugimura, Y. Endo, K. Shudo, and Y. Hitotsuyanagi, K. Yamaguchi, N. Aimi, K. Ogata, T. Kuramochi, H. Seki, R. Hata, H. Fujiki, M. Suganuma, T. Sugimura, Biochem. Biophys. Res. Commun., <u>90</u>, 976(1979).
  5. Y. Hitotsuyanagi, H. Fujiki, M. Suganuma, N. Aimi, S. Sakai, Y. Endo, K.

- 5. Y. Hitotsuyanagi, H. Fujiki, M. Suganuma, N. Aimi, S. Sakai, Y. Endo, K. Shudo, and T. Sugimura, Chem. Pharm. Bull., <u>32</u>, 4233(1984).
  6. a) S. Nakatsuka, H. Miyazaki, and T. Goto, Tetrahedron Lett., <u>21</u>, 2817(1980).
  b) S. Nakatsuka, H. Miyazaki, and T. Goto, Chemistry Lett., <u>102</u>(1000). 407(1981). c) S. Nakatsuka, K. Yamada, and T. Goto, Tetrahedron Lett., <u>27</u>, 4757(1986). d) S. Nakatsuka, O. Asano, and T. Goto, Heterocycles, <u>24</u>, 2109(1986). e) S. Nakatsuka, K. Ueda, O. Asano, and T. Goto, Heterocycles, <u>26</u>, 65(1987). f) S. Nakatsuka, O. Asano, K. Ueda, and T. Goto, Heterocycles, <u>26</u>, in press (1987). g) O. Asano, S. Nakatsuka, and T. Goto, Heterocycles, <u>26</u>, in press (1987). h) S. Nakatsuka, K. Teranishi, S. Hayashi, and T. Goto, Tetrahedron Lett., <u>28</u>, in press (1987).
- 7. S. Nakatsuka, T. Masuda, O. Asano, T. Teramae, and T. Goto, Tetrahedron Lett., 27, 4327(1986).
- 8. S. Nakatsuka, T. Masuda, K. Sakai, and T. Goto, <u>ibid.</u>, <u>27</u>, 5735(1986).
- 9. S. Nakatsuka, T. Masuda, and T. Goto, <u>ibid.</u>, <u>27</u>, 6245(1986).
- 10. Synthetic studies toward teleocidins by other groups were cited in reference 7.
- Synthetic studies on teleocidin III, T. Masuda, K. Ueda, O. Asano, S. Nakatsuka, and T. Goto, Heterocycles, 26, in press (1987).
   Typical proton signals of H-NMR spectra in CDCl, δ(ppm). 4(major isomer): 5.92(1H,br.m),7.59(1H,s),8.05(1H,s). 5g:0.73(3H,d,J=8),1.08(3H,d,J=8),1.28 5.92(1H, br.m), 7.59(1H, s), 8.05(1H, s). 53:0.73(3H, d, J=8), 1.08(3H, d, J=8), 1.28 (3H, s), 1.32(3H, s), 2.77(1H, d, J=16), 25:0.80(3H, d, J=8), 0.98 (3H, d, J=8), 1.28(3H, s), 1.52(3H, s), 2.85(2H, s). 6:2.20(2H, t, J=8), 3.36(1H, m), 3.68(1H, m), 7.22(1H, dd, J=3 and 2), 7.38(1H, t, J=3), 8.13(1H, s). 9a:3.79 and 3.83(3H, s), 3.93 & 3.94(1H, d, J=7). 9b:3.80 and 3.82(3H, s), 3.98 and 4.00(1H, d, J=6). 10a:3.73(3H, s), 5.12(1H, br.d, J=10), 5.40(1H, br.d, J=10). 12a:5.24 (1H, d, J=10), 5.38(1H, d, J=18), 6.14(1H, dd, J=18 and 10).
  13. For instance, treatment of 7 with formic acid at 25°C afforded a cyclized product similar to 2
- product similar to 2.

(Received in Japan 10 February 1987; accepted 19 May 1987)